

Procedures and Data

Heterocyclic Synthesis with Nitriles: Synthesis of some Novel Pyrrolo[2,1-*b*]thiadiazoline, Pyrrolo[2,1-*b*]thiadiazolo[3,2-*a*]pyrimidine and Pyridine Derivatives ¹⁾

Nadia H. Metwally, and Fathy M. Abdelrazek

Giza (A. R. Egypt), Chemistry Department, Faculty of Science, Cairo University

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Dedicated to Prof. Dr. N. A. Kassab on the Occasion of his 60th Birthday

Abstract. α -(Bromothiocyanatomethyl)benzylidene malononitrile (**1**) reacts with the hydrazides **2a–c**, cyanoacetohydrazide (**2d**), cyanoacetamide (**3a**) and cyanoacetanilides **3b–d** to afford pyrrolo[2,1-*b*]thiadiazolines **5a–c**, pyrrolo[2,1-

b]thia-diazolo[3,2-*a*]pyrimidine (**8**) and the pyridone derivatives **11a–d**. Structures and conceivable mechanisms are discussed.

Syntheses of pyrroles and fused pyrroles are known [1, 2]. We reported on approaches to pyrrole derivatives that can be used as biodegradable agrochemicals [3–7] and on the synthesis and reactions of α -(bromothiocyanatomethyl)benzylidene malononitrile (**1**) [8, 9]. Thiadiazoles fused to heterocyclic system possess also antimicrobial activity [10]. Now we describe further reactions of **1**.

In boiling ethanol in the presence of potassium carbonate **1** and hydrazides **2a–c** via assumed intermediates **4** (Scheme 1) yield pyrrolothiadiazoline imines **5a–c** and not, like **1** and phenylhydrazine alone [8], pyridazines **7**. In the IR spectra of **5a–c** no ν (SCN) was detectable, ¹H and ¹³C NMR data agree with the structures **5**. Similar cyclizations to the SCN to afford imino thiadiazoles are known [11]. In ethanolic HCl **5a–c** are easily transformed into **6a–c**, elemental analyses and spectral data proved the structures **6a–c**. If **1** reacts with cyanoacetohydrazide (**2d**), further cyclization of **5** ($R = COCH_2CN$) affords the pyrrolothiadiazolopyrimidine **8**; elemental analyses and spectral data agree with structure **8**.

The reaction of **1** with cyanoacetamide and cyanoacetanilides **3a–d** takes another route. Most likely amides **3** undergo a Michael type addition, paralleled or followed by a reverse reaction with loss of $BrCH_2SCN$ to afford the intermediate **10** (Scheme 2), and cyclization to the pyridone derivatives **11a–d**. The IR spectra of **11a–d** show two ν (CN) at 2202 and 2206 cm^{-1} and ν (CO) at 1660–1680 cm^{-1} ; elemental analyses and ¹H NMR spectra agree with the structures **11a–d**. Under the same conditions benzylidene malononitrile **9** reacts with cyanoacetamide **3a** to give a product identical with **11a** from **1**.

Experimental

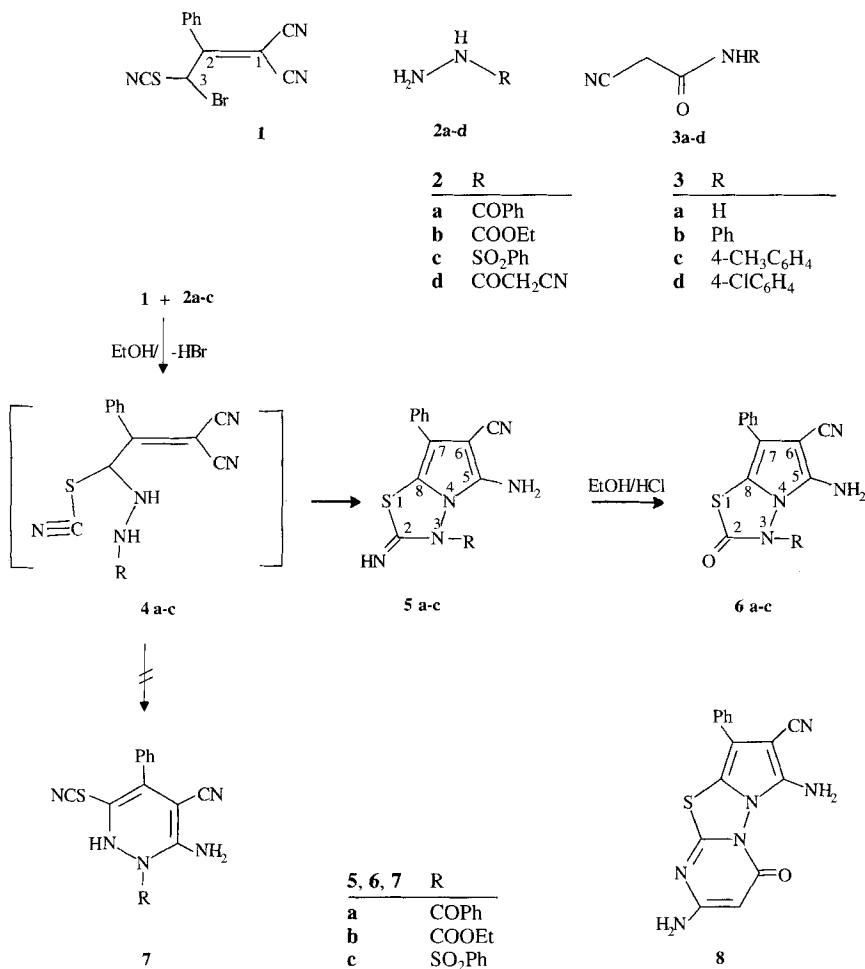
Melting points were determined on an electrothermal apparatus and are uncorrected. The IR spectra were recorded as KBr pellets on a Perkin Elmer 1430 spectrophotometer. The ¹H NMR and ¹³C NMR spectra were taken on a Varian Gemini 200 MHz spectrometer in DMSO-d₆ using TMS as internal standard. Mass spectra were taken on a Shimadzu GCMS-GB 1000 PX (70 ev). Elemental analyses were carried out by the Microanalytical Center at Cairo University.

The α -(bromothiocyanatomethyl)benzylidene malononitrile (**1**) was prepared according to the previously reported procedure [8]. – ¹³C NMR: δ /ppm = ([²H₆]DMSO) 133.66 (C-1), 131.91 (C-2), 128.50–129.69 (aromatic-C), 117.11 (CN), 116.23 (CN), 112.0 (SCN), 34.42 (C-3).

Reaction of α -(Bromothiocyanatomethyl)benzylidene malononitrile (**1**) with Hydrazides **2a–c**, Cyanoacetohydrazide (**2d**), Cyanoacetamide (**3a**) and Cyanoacetanilides **3b–d** (General Procedure)

To (3.04 g 0.01 mol) of **1** and 0.01 mol of each of **2a–c**, **2d**, **3a** or **3b–d** in ethanol (30 ml) was added a solution of potassium carbonate (1.38 g, 0.01 mol) in a minimum amount of water, and refluxed for 3–5 h (T. L. C control). The mixture was left to cool overnight, poured in cold water, and neutralized with HCl. The dark precipitates that appeared were separated by filtration and recrystallized.

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**Scheme 1**

5-Amino-3-benzoyl-6-cyano-7-phenylpyrrolo[2,1-b]thiadiazoline-2-imine (**5a**)

5a: 85% yield; *m.p.* > 330 °C (DMF/EtOH). – IR: ν/cm^{-1} = 3423–3200 (NH₂ and NH), 2206 (CN) and 1687 (hydrogen bonded C=O). – ¹H NMR ([²H₆] DMSO): δ/ppm = 7.2–8.0 (m, 10H), 8.18 (s, 2H) and 11.40 (s, 1H). – ¹³C NMR ([²H₆] DMSO): δ/ppm = 172.53 (CO), 165.3 (C-2), 156.34 (C-5), 145.04 (C-8), 134 (C-6), 126.24–133.4 (aromatic-C), 117.1 (CN) and 109.13 (C-7). – MS: *m/e* (%) 359.

C₁₉H₁₃N₅OS Calcd.: C 63.47 H 3.64 N 19.49 (359.22) Found: C 63.73 H 3.84 N 19.70.

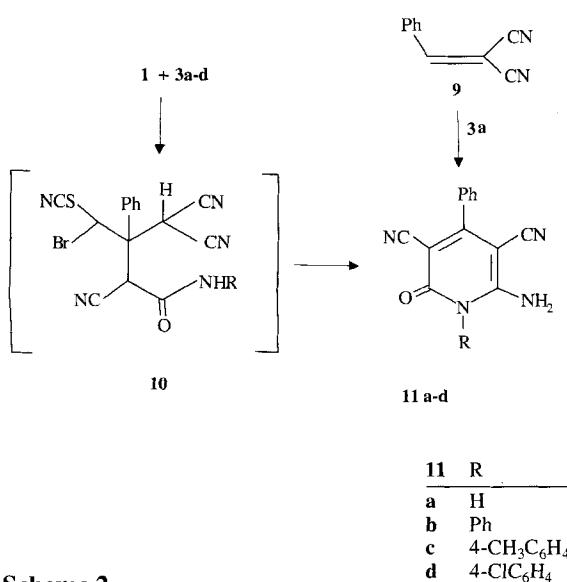
5-Amino-6-cyano-3-ethoxycarbonyl-7-phenylpyrrolo[2,1-b]thiadiazoline-2-imine (**5b**)

5b: 63% yield; *m.p.* > 300 °C (DMF/EtOH). – IR: ν/cm^{-1} = 3423–3200 (NH₂ and NH), 2205 (CN) and 1713 (hydrogen bonded ester C=O). – ¹H NMR ([²H₆] DMSO): δ/ppm = 1.3 (t, 3H), 2.41 (q, 2H), 7.2–7.81 (m, 5H), 8.25 (s, br., 2H) and 11.3 (s, 1H). – MS: *m/e* (%) 327.

C₁₅H₁₃N₅O₂S Calcd.: C 55.01 H 4.00 N 21.40 (327.21) Found: C 55.25 H 3.76 N 21.70.

5-Amino-3-benzenesulfonyl-6-cyano-7-phenylpyrrolo[2,1-b]thiadiazoline-2-imine (**5c**)

5c: 72% yield; *m.p.* > 300 °C (DMF/EtOH). – IR: ν/cm^{-1} = 3423–3200 (NH₂ and NH) and 2206 (CN). – ¹H NMR

**Scheme 2**

([$^2\text{H}_6$] DMSO): $\delta/\text{ppm} = 7.2\text{--}8.1$ (m, 10H), 8.31 (s, 2H) and 11.4 (s, 1H). – MS: m/e (%) 395.

$\text{C}_{18}\text{H}_{13}\text{N}_5\text{O}_2\text{S}_2$ Calcd.: C 54.64 H 3.31 N 17.72
(395.27) Found: C 54.91 H 3.55 N 17.50.

2,5-Diamino-6-cyano-7-phenylpyrrolo[2,1-*b*]thiadiazolo[3,2-*a*]pyrimidin-4-one (8)

8: 94% yield; *m.p.* > 300 °C (DMF/EtOH). – IR: $\nu/\text{cm}^{-1} = 3400\text{--}3230$ (NH_2), 2205 (CN) and 1650 (C=O). – ^1H NMR ([$^2\text{H}_6$] DMSO): $\delta/\text{ppm} = 7.2\text{--}7.7$ (m, 6H), 8.2 (s, 2H) and 8.4 (s, 2H). – MS: m/e (%) 322.

$\text{C}_{15}\text{H}_{10}\text{N}_6\text{OS}$ Calcd.: C 55.87 H 3.12 N 26.08
(322.19) Found: C 55.60 H 3.38 N 26.30.

2-Amino-3,5-dicyano-4-phenyl[1,6]dihydropyridin-6-one (11a)

11a: 86% yield; *m.p.* > 268 °C (DMF/EtOH). – IR: $\nu/\text{cm}^{-1} = 3400\text{--}3200$ (NH_2) and 2205 (CN), 2202 (CN) and 1680 (C=O). – ^1H NMR ([$^2\text{H}_6$] DMSO): $\delta/\text{ppm} = 7.21\text{--}8.2$ (m, 7H) and 10.4 (s, 1H). – MS: m/e (%) 236.

$\text{C}_{13}\text{H}_8\text{N}_4\text{O}$ Calcd.: C 66.07 H 3.41 N 23.72
(236.10) Found: C 66.37 H 3.11 N 23.60.

2-Amino-3,5-dicyano-1,4-diphenyl[1,6]dihydropyridin-6-one (11b)

11b: 95% yield; *m.p.* 190 °C (EtOH). – IR: $\nu/\text{cm}^{-1} = 3400\text{--}3200$ (NH_2) and 2206 (CN), 2204 (CN) and 1681 (C=O). – ^1H NMR ([$^2\text{H}_6$] DMSO): $\delta/\text{ppm} = 7.22\text{--}8.4$ (m, 12H). – MS: m/e (%) 312.

$\text{C}_{19}\text{H}_{12}\text{N}_4\text{O}$ Calcd.: C 73.05 H 3.87 N 17.94
(312.14) Found: C 73.25 H 3.67 N 18.20.

2-Amino-3,5-dicyano-4-phenyl-1-(4-methylphenyl)[1,6]dihydropyridin-6-one (11c)

11c: 97% yield; *m.p.* 205 °C (EtOH). – IR: $\nu/\text{cm}^{-1} = 3400\text{--}3200$ (NH_2) and 2206 (CN), 2204 (CN) and 1680 (C=O). – ^1H NMR ([$^2\text{H}_6$] DMSO): $\delta/\text{ppm} = 2.4$ (s, 3H), 7.1–8.22 (m, 11H). – MS: m/e (%) 326.

$\text{C}_{20}\text{H}_{14}\text{N}_4\text{O}$ Calcd.: C 73.59 H 4.32 N 17.17
(326.15) Found: C 73.74 H 4.12 N 17.30.

2-Amino-3,5-dicyano-1-(4-chlorophenyl)-4-phenyl[1,6]dihydropyridin-6-one (11d)

11d: 98% yield; *m.p.* 210 °C (EtOH). – IR: $\nu/\text{cm}^{-1} = 3400\text{--}3200$ (NH_2) and 2206 (CN), 2204 (CN) and 1683 (C=O). – ^1H NMR ([$^2\text{H}_6$] DMSO): $\delta/\text{ppm} = 7.1\text{--}8.3$ (m, 11H). – MS: m/e (%) 347.

$\text{C}_{19}\text{H}_{11}\text{N}_4\text{OCl}$ Calcd.: C 65.79 H 3.19 N 16.16
(346.58) Found: C 65.40 H 3.35 N 16.38.

Conversion of 5a–c into 6a–c

In ethanolic HCl (30 ml) was added 1 g of each of **5a–c**, the reaction mixture was refluxed for 4 hr, left to cool, then filtered off. The solid so obtained was crystallised:

5-Amino-3-benzoyl-6-cyano-7-phenylpyrrolo[2,1-*b*]thiadiazolin-2-one (6a)

6a: 80% yield; *m.p.* > 300 °C (DMF/EtOH). – IR: $\nu/\text{cm}^{-1} = 3400\text{--}3200$ (NH_2), 2206 (CN), 1687 (hydrogen bonded C=O) and 1650 (C=O). – ^1H NMR ([$^2\text{H}_6$] DMSO): $\delta/\text{ppm} = 7.2\text{--}8.1$ (m, 10H), 8.2 (s, 2H) and 11.5 (s, 1H). – ^{13}C NMR ([$^2\text{H}_6$] DMSO): $\delta/\text{ppm} = 175.2$ (C-2), 171.68 (CO), 156 (C-5), 147

(C-8), 135.2 (C-6), 126.05–133 (aromatic-C), 117.2 (CN) and 109.5 (C-7). – MS: m/e (%) 360.

$\text{C}_{19}\text{H}_{12}\text{N}_4\text{O}_2\text{S}$ Calcd.: C 63.30 H 3.35 N 15.55
(360.19) Found: C 63.50 H 3.18 N 15.70.

5-Amino-6-cyano-3-ethoxycarbonyl-7-phenylpyrrolo[2,1-*b*]thiadiazolin-2-one (6b)

6b: 60% yield; *m.p.* 290 °C (DMF/EtOH). – IR: $\nu/\text{cm}^{-1} = 3400\text{--}3200$ (NH_2), 2205 (CN) and 1713 (hydrogen bonded ester C=O), 1655 (C=O). – ^1H NMR ([$^2\text{H}_6$] DMSO): $\delta/\text{ppm} = 1.3$ (t, 3H), 2.4 (q, 2H), 7.2–7.82 (m, 5H), 8.31 (s, br., 2H) and 11.3 (s, 1H). – MS: m/e (%) 328.

$\text{C}_{15}\text{H}_{12}\text{N}_4\text{O}_3\text{S}$ Calcd.: C 54.85 H 3.68 N 17.07
(328.19) Found: C 54.60 H 3.82 N 17.30.

5-Amino-3-benzenesulfonyl-6-cyano-7-phenylpyrrolo[2,1-*b*]thiadiazolin-2-one (6c)

6c: 76% yield; *m.p.* 310 °C (DMF/EtOH). – IR: $\nu/\text{cm}^{-1} = 3400\text{--}3200$ (NH_2) and 2206 (CN), 1660 (C=O). – ^1H NMR ([$^2\text{H}_6$] DMSO): $\delta/\text{ppm} = 7.23\text{--}8.2$ (m, 10H), 8.41 (s, 2H) and 11.4 (s, 1H). – MS: m/e (%) 396.

$\text{C}_{18}\text{H}_{12}\text{N}_4\text{O}_3\text{S}_2$ Calcd.: C 54.51 H 3.05 N 14.13
(396.25) Found: C 54.70 H 3.31 N 14.4.

2-Amino-3,5-dicyano-4-phenyl[1,6]dihydro-pyridine-6-one (11a)

To **9** (0.01 mol) and **3a** (0.01 mol) in 30 ml of ethanol was added a solution of potassium carbonate (1.38 g, 0.01 mol) in a minimum amount of water and refluxed for 4 h. The mixture was left to cool overnight, poured on cold water and neutralized with HCl. The precipitate was recrystallized from DMF/EtOH: **11a**, yield, (1.2 g, 85%), *m.p.* 270 °C.

References

- [1] G. Baccolini, C. Sandali, J. Chem. Soc., Chem. Commun. **1987**, 788
- [2] E. Toja, A. Depaoli, G. Tuan, J. Kettenring, Synthesis **1987**, 272
- [3] F. M. Abdelrazek, A. A. Fadda, Z. Naturforsch., **1986**, 41B, 499
- [4] F. M. Abdelrazek, J. Prakt. Chem. **1990**, 479
- [5] F. M. Abdelrazek, A. M. Salah, Bull. Chem. Soc. Jpn. **1993**, 66, 1722
- [6] Z. E. Kandeel, A. M. Farag, F. M. Abdelrazek, Heteroatom Chem. **1995**, 6, 281
- [7] F. M. Abdelrazek, Z. E. Kandeel, A. M. Salah, Heteroatom Chem. **1995**, 1, 77
- [8] F. M. Abdelrazek, Heteroatom Chem. **1995**, 6, 211
- [9] F. M. Abdelrazek, M. S. Bahbouh, Phosphorous, Sulfur and Silicon, **1996**, 116, 235
- [10] N. F. Eweiss, A. A. Bahajaj, J. Heterocyclic Chem. **1987**, 24, 1173
- [11] A. O. Abdelhamid, A. S. Shawali, Z. Naturforsch., **1987**, 42B, 613

Address for correspondence:

Prof. Dr. Fathy Mohamed Abdelrazek
Chemistry Department, Faculty of Science
Cairo University
Giza
A. R. Egypt